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Toxicometabolomics of lindane in adult zebrafish (*Danio rerio*) using GC-MS/MS and LC-Orbitrap-MS/MS

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Abstract

Lindane is a broad-spectrum persistent organochlorine pesticide that has been used to control pests for many years. In this study, its toxic mechanisms in adult zebrafish were investigated using targeted metabolomics with GC-MS/MS and non-targeted metabolomics with LC-Orbitrap-MS/MS. Zebrafish was exposed to lindane in water for 48 h in three groups: control, low exposure ($1/10 LC_{50}$) and high exposure (LC_{50}). In the zebrafish exposed to low concentration of lindane, 2.24–3.98 mg/kg of lindane were determined, while 35.67–56.46 mg/kg were observed in the zebrafish exposed to high concentration. A total of 118 metabolites were identified from 394 metabolites on GC-MS/MS and 45 metabolites were selected as biomarkers. A total of 62 metabolites were identified on LC-Orbitrap-MS/MS and 7 metabolites were selected as biomarkers. Three groups were well separated on partial least squares-discriminant analysis (PLS-DA), and a total of 52 metabolites in both the targeted and non-targeted metabolites were selected as biomarkers, phane, phane, Five metabolic pathways such as the pentose phosphate pathway (PPP), histidine metabolism, phenylalanine metabolism, alanine/aspartate/glutamate metabolism, and phenylalanine/tyrosine/tryptophan biosynthesis, were observed to show toxicologically significant alterations. Oxidative stress was also confirmed through MDA and ROS assays. Such perturbations of the metabolic pathways of zebrafish caused by the exposure to lindane resulted in significant toxicological effects.

Keywords: Metabolomics, Lindane, Zebrafish, GC-MS/MS, LC-Orbitrap-MS/MS

Introduction

Chemicals called persistent organic pollutants (POPs) can cause environmental problems because they are predominant toxic contaminants with long half-lives to remain in the environment [1]. In particular, organochlorine pesticides POPs have carbon-chlorine bonds, which are very stable to hydrolysis and the many chlorine substituents and/or functional groups make them more resistance to biological degradation and photolysis [2]. It was noted that during the 1950s–1970s that

Republic of Korea

organochlorine pesticides such as lindane and Dichloro-Diphenyl-Trichloroethane (DDT) accounted for approximately 80% of total pesticides produced [3]. For this reason, organochlorine pesticides continue to cause environmental problems in many countries [4–6].

Lindane is a POP that is a broad-spectrum persistent organochlorine pesticide and has been used to control pests for many years [7]. It has inhibitory effects on gamma amino butyric acid (GABA) receptors in insects and vertebrates [8, 9]. The chloride channel blocking mechanism of lindane may involve non-competitive binding to channel-associated allosteric sites on the receptor. Toxic symptoms caused by lindane include spontaneous increases in nerve activity and convulsive seizures in insects and vertebrates [10]. Lindane produced various metabolites in bio-environmental systems



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such as hexachorocyclohexene, pentachlorocyclohexene, tetrachlorobenzenes, trichlorobenzenes, dichlorobenzene, tetrachlorophenol, trichlorophenols, dichlorophenol and dichlorophenylglutathiones and etc. [11]. The US Environmental Protection Agency banned lindane use in August 2006, but the environmental problems associated with lindane remains [12]. In the environment, lindane is more soluble in water than most other organochlorine compounds therefore it has a greater possibility of remaining in the water column [13, 14]. Because of its strong hydrophobic property and long persistency, lindane can bioaccumulate in fish, microorganisms, and mammals [15, 16]. Therefore, it is necessary to conduct research on lindane exposure in aquatic ecosystems. Zebrafish are known as model organisms for studies of physiology, vertebrate biology, and human disease [17]. In a human drug conversion study, about 86% of the 1318 human drug targets have been identified in zebrafish [18]. Many previous metabolomics studies have been conducted with zebrafish because of their special feature like the genetic homology with mammals, prolificacy, portable size, and economics [19-23].

From the 1990s, metabolomics was defined as post genomics area and is the closest to the phenotype in the "omics cascade" to be the end point of that cascade. It aimed to identify the changes of biochemical pathways of animals, plants and microbes in response to physiological stimuli or genetic disturbances. Therefore, measuring metabolites present in cells, tissues or organisms is a key task in metabolomics research, however, there is currently no single technological platform to analyze all metabolites [24, 25]. In metabolomics analytical techniques, the most commonly used analytical methods are those using GC-MS, LC-MS, and NMR [26]. Each of these technologies has its own unique advantages and disadvantages. Many previous studies have been conducted to confirm the biological responses to environmental stress using NMR [21, 27, 28]. The main disadvantage of NMR is that it is a relatively low-sensitivity technique that has a high detection limit of about $1-5 \,\mu M$ and requires a relatively large sample amount (~500 μ L) [29]. Compared with NMR, MS can separate many metabolites with high resolution and sensitivity [30]. GC-MS-based metabolomics can cover a certain polarity range [31]. In addition, GC-MS is an ideal analytical equipment for the analysis of both volatile and non-volatile compounds through derivatization of non-volatile polar compounds. The latest capillary GC is an excellent tool for the analysis of complex metabolic mixtures, providing the chromatographic separations with high resolution and reproducibility [30]. Moreover, compared to LC-MS based analysis, GC-MS has the advantage of much greater resolution, and has a large spectral library,

and can separate small compounds that tend to be eluted to the solvent front early in conventional reverse-phase LC-MS analysis into good retention [32]. Therefore, in targeted metabolomics study, GC-MS has the advantage of being able to accurately and quantitatively detect more diverse metabolites than LC-MS. However, LC-Orbitrap-MS-based metabolomics provides positive and negative ionization, which increases the variety of metabolites that can be detected [31, 33]. These key properties have led to an increase in the use of Ultra-high performance liquid chromatography tandem mass spectrometry (UHPLC-MS/MS) related procedures in recent years [34, 35]. The strategies used for metabolomics experiments include 'targeted' and 'non-targeted' metabolomics [36]. Targeted metabolomics uses multiple analyses of defined metabolites that are characterized chemically and annotated biochemically [24]. On the other hand, non-targeted metabolomics uses a comprehensive method to analyze unknown metabolites [37].

In this study, both GC-MS/MS-based targeted metabolomics and LC-Orbitrap-MS/MS-based non-targeted metabolomics were performed after zebrafish exposure to lindane. A total of 394 metabolomes were analyzed by GC-MS/MS with multiple reaction monitoring (MRM) mode in targeted metabolomics, and LC-Orbitrap-MS/ MS with scan mode was used in non-targeted metabolomics to identify metabolites in zebrafish. Statistical analyses were performed by SIMCA⁺ and MetaboAnalyst 5.0 to identify biomarker metabolomes and obtain a heatmap. The metabolic pathways were identified by MetaboAnalyst 5.0 to investigate the toxicological and biochemical effects of lindane in zebrafish.

Materials and methods

Chemicals and reagents

Analytical-standard lindane (98.5%) was purchased from Dr. Ehrenstorfer GmbH (Augsburg, Germany), and ribitol, as an internal standard, was purchased from Wako. TMCS-MSTFA reagent (1%) was purchased from Thermo Fisher Scientific Company. Methoxyamine hydrochloride, N,N-dimethylformamide (DMF), and pyridine were purchased from Sigma-Aldrich. Acetonitrile and water were of LC-MS grade and were purchased from Fisher Scientific. Acetone and methanol were purchased from Merck (Darmstadt, Germany) and were of LC-MS grade. Reagents with the highest available grades were used. The methoxyamination reagent was prepared by dissolving 20 mg of methoxyamine hydrochloride in 1 mL of pyridine.

Experimental animals and chemical exposure

Chemical exposure were carried out as previously described [38] with slight modification. Adult zebrafish

with the body weight of 400–500 mg and the length of 3–4 cm were used, and gender was not considered. Adult zebrafish were acclimated at 23 ± 1 °C in a glass water tank for 1 week. The photoperiod was maintained as a 16:8 h light:dark cycle. During this period, zebrafish were fed commercial fish feed, and were not fed on the day before lindane exposure. The LC₅₀ value of lindane for zebrafish is 110 µg/L for 48 h [39]. Based on this value, the exposure concentrations were set to 110 µg/L (1100 µg/mL in acetone, 0.5 mL) and 11 µg/L (110 µg/mL in acetone, 0.5 mL) for the high-exposure and low-exposure groups, respectively.

In 10 L beakers, 20 zebrafish per group were kept for 48 h before exposure to 110 μ g/L (high exposure; LC₅₀) and 11 μ g/L (low exposure; 1/10 LC₅₀) of lindane, including a control group without lindane. After exposure, 6 zebrafish in each group were sampled for metabolome and oxidative stress analyses. During the experiment, the exposure solution was changed daily to maintain the desired lindane concentration.

Metabolomic sample preparation, profiling and metabolome identification by GC-MS/MS

Sample preparation procedure was performed as previously described [38]. Fish samples were frozen in liquid N_2 in 5 mL Eppendorf tubes with 3 of 3 mm and 2 of 7 mm Retsch stainless steel were ground by shaking vigorously using a MiniG (SPEX Sample Prep, USA). One mL of 50% MeOH solution containing 0.1 µg/ mL of ribitol (internal standard) was added to a 50 mg of ground sample in a 2 mL tube, and the sample was then vortexed at 4 °C for 10 min. After centrifuging the extracts for 10 min at 13,000 rpm, 100 µL of supernatants of the extracts were evaporated by a speed vacuum concentrator (Hanil Modulspin 40). The dried residues were reacted with 50 µL of methoxyamination reagent for 90 min at 37 °C with shaking and were then derivatized with 50 µL of MSTFA reagent (MSTFA+1% TMCS) for 30 min at 37 °C.

For metabolite analysis, a Shimadzu GCMS-TQ8040 equipped with a BPX-5 column (30 m \times 0.25 mm i.d., 0.25 µm film thickness, TRAJAN) was used in MRM mode and a total of 394 metabolites (332 metabolites from the Smart Metabolites Database, and 62 metabolites from an in-house library) were analyzed. The injection volume was 1.0 µL (split mode; 30:1), and the temperatures for the injector, transfer lines and ion source (70 eV) were 250 °C, 280 °C and 200 °C, respectively. The initial oven temperature was 60 °C [2 min], and the temperature was ramped to 320 °C at a rate of 10 °C/min and then maintained for 15 min. Helium was used as the carrier gas with a flow rate of 1 mL/min and argon was

used as the collision gas. Data processing was performed with manual reconfirmation of the peak detections using GCMS Solution software (version 4.3, Shimadzu).

Non-targeted metabolite profiling and identification by LC-Orbitrap-MS/MS

Non-targeted metabolite profiling and identification were conducted as previously described [40] with slight modification. Fish extract (5 μ L) before derivatization from the targeted metabolomics method was used for non-targeted metabolite analysis, which employed a Q-Exactive Orbitrap mass spectrometer (Thermo Fisher Scientific Inc., Waltham, MA) that was coupled with a UHPLC (DGU-20A) system. The separations were performed at 50°C with an HSS T3 C18 column (2.1×100 mm, 1.8μ m, Advanced Materials Technology, Wilmington, DE). The flows were constant at 0.3 mL/min with 0.1% formic acid in water (A) and 0.1% formic acid in methanol (B) as the mobile phases. The following gradient was used: 10% B for 0-0.2 min, 10% to 55% for 0.2-15 min, 55% to 98% for 15-20 min, 98% B for 20-35 min, 98% to 10% B for 35-35.1 min and 10% B for 35.1-40 min.

Mass spectra were obtained with an HESI (heated electrospray ionization) source that was operated in positive and negative modes. The parameters for the HESI source were set as follows: the sheath gas flow rate was 35 (arbitrary units), auxiliary gas flow rate was 10 (arbitrary units), spray voltage was 4 kV, and heater temperature was 350°C. The MS was set to full scan with ddMS² from m/z 100 to 1500. In the data interpretation process, the mass error range was set to 5 ppm or less, and the metabolites were identified to have a mzCloud matching score of 80% or more.

Determination of lindane concentration in zebrafish using GC-MS/MS

Lindane was extracted from 200 µL of fish extract before derivatization with 200 µL of hexane. One µL of hexane layer was analyzed with GC-MS/MS. Calibration curve solutions (10, 20, 50, 100 and 200 µg/L) were prepared with control fish extract to compensate matrix effect, and they were also extracted with hexane before GC-MS/MS analysis. Analytical condition of GC-MS/MS was identical with metabolites analysis as described above with specific MRM mode for lindane (quantification: m/z181.0 m \rightarrow/z 145.0, and qualification: m/z 219.0 m \rightarrow/z 183.0). Samples exposed to high concentration of lindane were diluted by 10–20-fold with hexane for quantitation.

Statistical analysis of metabolomes and metabolic pathway analysis

Multivariate data analyses were performed using SIMCA-P⁺ software (version 12.0.1, Umetrics, Sweden). The supervised multidimensional statistical model, which is a partial least squares-discriminant analysis (PLS-DA), was conducted to differentiate between each treatment group. By confirming the VIP (variable importance in the projection) scores in PLS-DA, metabolites with VIP scores > 1 and standard errors < 1 were selected. The p-value of selected metabolites were calculated using one-way ANOVA, and metabolites with p<0.05 were selected as biomarker metabolites affected significantly by lindane exposure. A heatmap of selected biomarker metabolites was generated using MetaboAnalyst 5.0 (www.metaboanalyst.ca). MetaboAnalyst 5.0 also produced the metabolic pathway analysis plots based on the Danio rerio KEGG library.

Measurement of ROS and MDA contents

ROS and MDA contents were measured according to manuals with commercially available kits, respectively. The contents of reactive oxygen species (ROS) were measured by DCFH-DiOxyQ (OxiSelectTM In vitro ROS/ RNS assay kit, Cell Biolabs, Inc., CA, USA). After adding 500 µL of phosphate buffered saline (PBS) to a 20 mg of powder sample, it was homogenized and centrifuged at 10,000 g for 5 min. Fifty microliters of the supernatant was mixed with 50 µL of catalyst and incubated at room temperature for 5 min. DCFH solution (100 µL) was added to the incubated mixture before further incubation for 40 min at room temperature. Fluorescence intensities of incubated samples were measured using a spectrophotometer (SpectraMax i3, Molecular Devices) at 480 nm excitation/530 nm emission.

The contents of malondialdehyde (MDA) were determined using a thiobarbituric acid (TBA) assay (OxiSelect[™] TBARS assay kit, Cell Biolabs, Inc., CA, USA). Into 30 mg of sample powder, 500 µL of 1X butylated hydroxytoluene (BHT) solution was added and it was centrifuged at 10,000 g for 5 min after homogenization. One hundred microliters of the supernatant was mixed with 100 µL of SDS lysis solution and incubated at room temperature for 5 min. A 250 µL of TBA reagent was added to the incubated mixture and further incubated for 60 min at 95 °C. After the incubation was completed, the samples were cooled to room temperature and were centrifuged at 3000 rpm for 15 min. To prevent interference with hemoglobin and its derivatives, 300 μ L of the supernatant was extracted with 300 µL of butanol. The fluorescence intensities of the samples in the butanol layer were measured using a spectrophotometer (SpectraMax i3, Molecular Devices) at 540 nm excitation/590 nm emission.

Results and discussion

Determination of lindane concentrations in zebrafish using GC-MS/MS

Lindane was extracted from fish extract with hexane [41] and analyzed with GC-MS/MS. Retention time of lindane was 16.8 min. Linearity of calibration curve (r2) was>0.99, indicating good correlation. In the zebrafish exposed with low concentration of lindane, 2.24–3.98 mg/kg of lindane were determined, while 35.67–56.46 mg/kg were observed in the zebrafish exposed with high concentration (Table 1).

Metabolome profiling using GC-MS/MS and LC-Orbitrap-MS/MS

Metabolomics analysis was performed using the 'targeted' and 'non-targeted' methods on GC-MS/MS and LC-Orbitrap-MS/MS. Recently, many metabolomics studies have been conducted using GC-MS and LC-MS, which are complementary mass spectrometry techniques. These techniques can identify an extensive range of metabolites with good sensitivity and selectivity and can provide complementary data to save experimental time with a single sample preparation process [42–44]. In targeted metabolomics, 394 metabolites (332 metabolites were from the Smart Metabolites Database and 62 metabolites were from our own laboratory library) were analyzed using GC-MS/ MS with MRM mode. These metabolites included 78 carbohydrates, 41 fatty acids, 18 alcohols, 75 amino acids, 9 nucleosides, 102 organic acids, 10 amines, 7 pyridines, 4 indoles, 11 purines, 2 nucleotides, 10 amides, 2 tocopherols, 6 steroids, 9 phenols, 4 pyrimidines, 3 esters, 1 inorganic compound, 2 isoprenoids, 1 glyceride, and 1 azole. After the GC-MS/MS analyses, manual identification of peaks was performed by confirming the corresponding metabolite MRM ions, peak retention times and areas. In these analyses, 118 confirmed metabolites were detected in whole zebrafish bodies (Fig. 1).

In non-targeted metabolomics, a total of 62 metabolites were detected and identified after LC-Orbitrap-MS/MS analysis and data processing with Compound Discoverer 3.1 (CD3.1). These metabolites included the following: 7 amines, 14 amino acids, 7 carbohydrates, 9 fatty acids, 2 fatty acyls, 3 steroids, 3 purines, 7 organic acids, 2 glycerides, 1 alcohol, 1 indole, 1 polyethylene glycol, 1 pterin, 1 pyridine, 1 stilbene, 1 aniline, and 1 organosulfonic acid (Fig. 2).

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Sample	1	2	3	Average
Low exposure	2.24 mg/kg	3.98 mg/kg	3.79 mg/kg	3.34 mg/kg
High exposure	35.67 mg/kg	42.32 mg/kg	56.46 mg/kg	44.81 mg/kg



Metabolomic alteration induced by lindane exposure

According to the PLS-DA plot of the targeted metabolomics data, the control group and 2 treated groups showed significant differences in their metabolic profiles (Fig. 3). VIP analyses of the PLS-DA model and p-value analyses of the ANOVA model were conducted to obtain 45 metabolites as biomarkers that contributed significantly to differentiating the metabolic profiles between the control and the two exposed groups (Table 2). From the same statistical analysis using non-targeted metabolomics data, 7 additional biomarker metabolites were identified. Table 3 shows the list of biomarker metabolites that were finally identified from the non-targeted method, including their molecular formulas, adducts, molecular weights, retention times, mzCloud matching scores and HMDB numbers. Therefore, a total of 52 biomarkers were found as a result of the statistical analyses of targeted and non-targeted metabolome data. The heatmap of these biomarkers clearly shows the pattern of metabolite changes (Fig. 4). The colors in the heatmap represent the metabolite levels from highest (red) to lowest (blue), which demonstrate a clear difference between the control and treatment groups.

Metabolic pathway and function analysis

Metabolic pathway analysis was carried out using MetaboAnalyst 5.0 with the Kyoto Encyclopedia of Gene and Genomes (KEGG). Five main perturbed pathways were identified, including the pentose phosphate pathway





(PPP), histidine metabolism, phenylalanine metabolism, alanine/aspartate/glutamate metabolism, and phenylalanine/tyrosine/tryptophan biosynthesis (Fig. 5). The metabolic pathways that were disturbed by lindane exposure and the trends of the related metabolites are summarized in Fig. 6.

In the PPP, the levels of ribose, ribose 5-phosphate, gluconic acid, gluconic acid lactone, 6-phophogluconic acid, glyceric acid, and sedoheptulose 7-phosphate were downregulated in zebrafish after lindane exposure. The decreases in these metabolites are probably caused by downregulation of glucose-6-phosphate dehydrogenase (G6PD) gene expression after lindane exposure [45]. G6PD plays a role in the oxidative step of the PPP and catalyzes the rate-limiting step to produce NADPH; therefore, the expression and activity of the G6PD-related gene tightly regulates PPP [46]. In addition, in the non-oxidation step of PPP, erythros-4-phosphate, which can contribute to the formation of aromatic amino acids such as phenylalanine, tryptophan, histidine and tyrosine, is produced, so the disturbance of PPP will affect the formation of amino acids [47].

With respect to phenylalanine/tyrosine/tryptophan biosynthesis and phenylalanine metabolism, phenylalanine and tyrosine decreased. The downregulation of these metabolites may lead to a negative impact on the synthesis of important neurotransmitters and hormones such as dopamine, epinephrine and norepinephrine [48]. Interruption of phenylalanine/tyrosine/tryptophan biosynthesis has an effect on the nervous system [19]. Moreover, an unstable phenylalanine metabolic state negatively affects the body's normal growth, development and physiological functions [49]. In a metabolomics study of zebrafish embryos that were exposed to fluoxetine, phenylalanine and tyrosine were downregulated in embryos [19].

In the present study, the relative amounts of the corresponding metabolites associated with alanine/aspartate and glutamate metabolism (alanine, glutamate and aspartate), were all downregulated. A study of metabolomics in medaka reported that alanine and tyrosine were downregulated after medaka embryos were exposed to dinoseb [50]. The downregulation of alanine and tyrosine resulted from energy expenditure [19, 50]. In addition, alanine is an essential nitrogenous product that is produced by the intestinal catabolism of glutamine, glutamate, and aspartate [51]. Downregulation of glutamate and aspartate may negatively affect the role of alanine in transporting some dietary amino acids to intestinal tissues [52].

In histidine metabolism, the concentrations of histidine, glutamate, and aspartate were downregulated. A study that evaluated the antitumor effects on A549 cells reported that histidine plays a role in inhibiting cell proliferation and migration in lung cancer cells and has antitumor effects [53]. In addition, a metabolomics study of bovine colostrum and mature milk reported that brain and nervous system development was encouraged by histidine in newborns [54]. Therefore, a decrease in histidine may increase the risk of cancer and have devastating effects on the brain and nervous system. Moreover, glutamate, a precursor of

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Š.	Compound name	VIP	Standard error	P value No	0	compound name	VIP S	tandard error	P value	No.	Compound name	VIP Stand	ard error P value	
<i>—</i>	Hypoxanthine	1.35	0.49	1.41E-06 16	б Э	-Aminoglutaric acid	1.19 0	39	3.69E—04	31	Ribulose 5-phosphate	1.23 0.62	5.52E-03	~
2	2-Aminoethanol	1.38	0.55	1.44E-06 17	P	almitic acid	1.31 0	29	4.73E—04	32	Docosahexaenoic acid	1.03 0.59	6.42E-03	~
e	Lysine	1.31	0.28	7.29E-06 18	۵ ۵	roline	1.16 0	39	5.91E04	33	Arachidonic acid	1.01 0.58	7.45E-03	~
4	Tyramine	1.31	0.28	9.58E-06 19	9	eucine	1.15 0	39	6.81E04	34	Eicosapentaenoic acid	1.00 0.57	7.52E-03	~
Ś	Xanthine	1.26	0.46	2.69E-05 20	<	Aethionine	1.13 0	39	9.15E—04	35	6 — Phosphogluconic acid	1.00 0.48	7.90E-03	~
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~	Mannose 6-phosphate	1.28	0.29	3.92E-05 22	2 T	yrosine	1.10 0	.45	1.71E03	37	Myristic acid	1.02 0.51	8.62E-03	~
œ	Glucose	1.28	0.36	5.09E-05 23	е С	libose	1.11 0	71	2.25E-03	38	Heptadecanoic acid	1.44 0.57	9.43E-03	~
6	Glucose 6-phosphate	1.26	0.31	7.23E-05 24	4	Jleic acid	1.09 0	52	2.62E-03	39	Ribose 5—phosphate	1.28 0.77	1.55E-02	
10	gluconic acid lactone	1.26	0.39	1.01E-04 25	ы У	laidic acid	1.08 0	53	2.70E-03	40	Stearic acid	1.32 0.57	2.18E-02	
]]	Valine	1.23	0.26	1.19E-04 26	6	ilyceric acid	1.15 0	53	2.73E-03	41	Glutamic acid	1.41 0.96	2.51E-02	
12	Sedoheptulose 7-phos- phate	1.22	0.51	1.40E—04 27	0	creatinine	1.21 0	64	2.95E—03	42	Methyl serine	1.51 0.78	2.55E-02	
13	Isoleucine	1.19	0.39	2.84E-04 28	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	ryptophan	1.05 0	58	3.00E-03	43	Threonine	1.00 0.64	2.58E-02	
4	Aspartic acid	1.21	0.39	3.31E-04 29	⊲	Vanine	1.07 0	.43	4.40E—03	4	Caproic acid	1.58 0.66	2.94E-02	
15	Phenylalanine	1.18	0.36	3.51E-04 30	ر 0	Jracil	1.05 0	.49	5.34E—03	45	Histidine	1.13 0.52	3.12E-02	
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Table 2 Significantly altered metabolites of zebrafish exposed to lindane in targeted metabolomics (VIP > 1.0, SE < 1.0, and p-value < 0.05)

No	Metabolite	Molecular formula	Adduct	Molecular weight	RT (min)	mzCloud matching score (%)	HMDB	VIP	Standard error	p-value
_	5(S), 15(S)-DiHETE	C ₂₀ H ₃₂ O ₄	-[HW]	336.23	20.60	95.30	HMDB0010216	1.95	0.78	8.0E-05
2	Lipoxin B4	C ₂₀ H ₃₂ O ₅	-[HW]	352.22	19.92	84.00	HMDB0005082	1.94	0.47	1.1E-04
ć	2-(14,15-Epoxyeicosatrienoyl) glycerol	C ₂₃ H ₃₈ O ₅	[M+H] ⁺	376.26	22.13	85.60	HMDB0013651	1.66	0.76	4.1E-03
4	Drospirenone	C ₂₄ H ₃₀ O ₃	[M+H] ⁺	366.21	21.76	89.30	HMDB0015467	1.51	0.59	1.5E-02
5	Choline	C ₅ H ₁₄ NO	[M+H] ⁺	103.10	1.28	97.30	HMDB0000097	1.49	0.31	1.8E-02
9	Adrenic acid	C ₂₂ H ₃₆ O ₂	[M+H] ⁺	332.27	23.59	94.10	HMDB0002226	1.44	0.78	2.3E-02
7	N6,N6,N6-Trimethyl-L-lysine	$C_9H_{20}N_2O_2$	[M+H] ⁺	188.1	1.20	86.00	HMDB0001325	1.36	0.76	3.0E-02

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GABA, which is involved in both alanine/aspartate and glutamate metabolism and histidine metabolism was decreased in the lindane exposer groups. Lindane, a neurotoxin, is known as an inhibitor of GABA-gated chloride channel inducing partial repolarization of neurons, resulting in involuntary muscle contraction and seizures [10]. Therefore, decreasing glutamate demonstrated that neurotransmitter-mediated nervous system was unable to be regulated.



Oxidative stress induced by lindane

To confirm the oxidative stress from exposure to lindane, ROS and MDA assay tests were conducted with zebrafish. The amounts of ROS were detected by DCFH-DA assays. Excessive oxidative stress can lead to the accumulation of ROS, which will damage major cell components [55]. Higher ROS production levels may affect the oxidation of proteins and lipids, disturbance of gene expression, and disruption of cellular redox states [56]. In addition, the accumulation of ROS lead to cell apoptosis [57]. In the study of apoptosis in thyroid cancer cell, inhibition of the PPP cause ROS-induced apoptosis in thyroid cancer cells [58]. In this study, PPP was downregulated in the pathway analysis and ROS content increased, therefore, apoptosis could have been induced. MDA, one of the major products of lipid peroxidation in cell membranes, is used as an index of lipid oxidation [59]. As a result, ROS and MDA



respectively



in the exposure group under high lindane concentrations were significantly increased (p < 0.05) (Fig. 7), which suggested oxidative damage to the cells.

Authors' contributions

XY conducted the sample preparation and sample analysis and wrote the manuscript. JL and EP helped with the sample preparation and analysis. H-KL revised the final manuscript. JHK supervised the project and revised the final manuscript. All authors read and approved the final manuscript.

Availability of data and materials

All data generated or analyzed during this study are included in this published article.

Declarations

Competing interests

The authors declare that they have no competing interests.

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