



이학박사 학위논문

Proton Beam Irradiation Effect on Nanomaterials: WSe₂ and MAPbI₃

고에너지 양성자 조사가 나노 물질(이차원 텅스텐 다이셀레나이드와 메틸암모늄 납 할라이드 페로브스카이트)에 미치는 영향

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물리·천문학부

신 지 원

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고에너지 양성자 조사가 나노 물질(이차원 텅스텐

다이셀레나이드와 메틸암모늄 납 할라이드

페로브스카이트)에 미치는 영향

지도 교수 이 탁 희

이 논문을 이학박사 학위논문으로 제출함 2022 년 8 월

서울대학교 대학원

물리·천문학부 물리학 전공 신 지 원

신지원의 이학박사 학위논문을 인준함 2022 년 8 월

위 위	원 장	차 국 린	(인)
부위원장		이 탁 희	(인)
위	원	민 홍 기	(인)
위	원	이 지 은	(인)
위	원	조 경 준	(인)

Abstract

Proton Beam Irradiation Effect on Nanomaterials: WSe₂ and MAPbI₃

Jiwon Shin

Department of Physics and Astronomy Seoul National University

The study of the effect of proton beam irradiation on electronic devices such as transistors and solar cells has several advantages. The first is that the electrical or optical performance of a device can be controlled. A high-energy proton beam can be used to induce defects in the device to control the electrical characteristics such as current level, and a low-energy proton beam can be used to exfoliate the layered material into a monolayer to significantly improve its optical properties. However, further studies are needed to elucidate the mechanism of this phenomenon. The second is that it is possible to explore the potential of using the electronic devices in high-radiation environments such as space. Especially, research of the proton irradiation effect is important in terms of space application because protons occupy a high proportion among various elements of cosmic ray such as gamma rays, electrons, and neutrons. Organic-inorganic lead halide perovskite, one of the materials I researched, is attracting attention as a nextgeneration solar cell material due to its high power conversion efficiency. Also, it has high potential for the space industry because of its high radiation hardness. Due to these advantages, it is being actively studied, but more research is needed to take full advantage of their stability and performance under radiation conditions.

In this manner, first I fabricated WSe₂ (one of transition metal chalcogenide) ambipolar field effect transistors and investigated the proton irradiation effect. Transition metal chalcogenide is a two-dimensional material attracting attention as a next-generation device material because of its remarkable electrical and optical properties such as high mobility and high photosensitivity. Among them, WSe₂ has a unique property that the carrier type of the electronic device varies depending on the thickness; p-type for thin thickness, ambipolar for intermediate thickness, and n-type for thick. Here, an ambipolar type transistor was designed and fabricated to examine the proton effect on both the electron accumulation region and the hole accumulation region. The electrical and physicochemical properties of the devices were measured and systematically compared before and after the high energy proton beam irradiation of 10 MeV under various dose conditions. The physicochemical properties did not change within the measurement limits. However, the electrical characteristics such as current level and threshold voltages were changed after the irradiation and the amount (and direction) of change were affected by the dose. These changes were explained by the proton irradiationinduced traps referred to herein as "positive bulk traps" and "negative interface traps".

Secondly, I synthesized organic-inorganic lead halide perovskite by two different methods, mechanochemical synthesis and flash evaporation, and

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investigated the proton irradiation hardness of the synthesized perovskite. Since both methods are dry synthesis methods, they have the advantage of being less toxic than the spin coating method, which is one of the representative perovskite synthesis methods. The synthesized materials were exposed to high-energy proton beams under various irradiation doses and physicochemical properties of the materials were compared before and after the proton irradiation. The properties of mechanochemically synthesized perovskite did not change noticeably even under high irradiation doses, whereas the properties of flash-evaporated perovskite did. This difference was explained by the stronger bonding energy due to its less defects in mechanochemically synthesized perovskite.

주요어 : Proton irradiation, Nanomaterial, Transition metal dichalcogenide, Tungsten diselenide, Organic-inorganic perovskite, Mechanochemical synthesis 학 번:2016-20297

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Chapter 1. Introduction

1.1. Proton matter interaction

1.1.1 Energy loss process of irradiated charged particle

When charged particle such as a proton is irradiated to a target material, the proton loses energy by interacting with the material, and the process can be divided into two steps. First step is the electronic stopping process, which initially dominates until the protons are slowed down sufficiently. The other step is the nuclear stopping process, which dominates near the stopping depth, where the protons have been sufficiently decelerated. Electronic stopping process is an ionizing process in which incident protons create electron-hole pairs in the target material. In this process, protons lose energy homogeneously, causing the degradation on the target material such as a change in the threshold voltage. On the other hand, the nuclear stopping process includes all non-ionizing processes that damages the lattice structure of the target material. In this process, protons lose their energy rapidly and causes degradation of materials such as cracks in the oxide layer. Since the energy loss mechanism of irradiated particles is affected by the thickness of the target material, the stopping depth of the protons is one of the important factors in proton effect studies. The stopping depth is determined by the stopping power that is affected by factors such as the type of target material and the energy of the proton. We will discuss this in the next part.

1.1.2 Stopping power

Stopping power, which is the loss energy per unit path length of irradiated charged particles in a target material, can be expressed as a formula and the best known is the Bethe equation.

$$S(T) = -\frac{dT}{dx} = n_{ion}\overline{I} = \frac{4\pi z^2 e^2 nZ}{m\beta^2 c^2} \left[\ln\left(\frac{2mc^2\gamma^2\beta^2}{\overline{I}}\right) - \beta^2 \right]$$

Here, T is the kinetic energy of the irradiated charged particle, n_{ion} is the number of ionized atoms in target material, \bar{I} is the average ionization energy, m is the mass of particle in the material that interact with irradiated particle, z is the charge number of incident particle (1 for proton), Z is the atomic number of target material, γ is the lorentz factor and β is the speed of the irradiated particle relative to the speed of light. Stopping power can be determined theoretically and experimentally with an accuracy of several % in the energy range of $0.1 \leq \beta \gamma \leq 1000$.

As can be seen from the Bethe equation, the stopping power increases as the speed of the proton decrease. While the irradiated particle travels through the material, its speed decreases and the stopping power increases toward the end of range. Shortly before the stopping depth where energy drops to zero, the force increases rapidly and reaches a maximum called as the Bragg peak. This tendency is also consistent with the simulation results in figure 1.1.

We simulated energy-loss profiles using Stopping and Range of Ions in Matter (SRIM) software. Figure 1.1 shows the SRIM results under various irradiated proton energy of 1 MeV, 10 MeV, 100 MeV and 1 GeV. While the energy

of proton (which is related to the speed of the proton) increases by 10 to the third power from 1 MeV to 1 GeV, the stopping depth increases by 10 to the fifth power from 14.6 μ m to 1.65 m. Also, the maximum value of energy lose per angstrom decreased from 12 to 0.18 eV/Å, as the proton energy increased. This show the stopping power decreases as the irradiated proton velocity increases.



Figure 1.1 SRIM results of various energy of proton beams

Other important factor influencing the stopping power is the type of target material. As the Beth equation shows, the stopping power is proportional to the atomic number. Figure 1.2 shows the SRIM results for various target materials

from beryllium (with atomic numbers 4) to lead (with atomic number 82) with proton energy of 10 MeV. As the atomic number increased from 4 to 82, the stopping depth decreased from 804 μ m to 301 μ m. This shows that the stopping power increases as the atomic number increases. Proton irradiation effect studies are being actively conducted by controlling these factors such as proton energy, and type and depth of target material according to the purpose of the study.



Figure 1.2 SRIM results of various materials with atomic number from 4 to 82

1.2. Transition metal dichalcogenide and tungsten diselenide

Two-dimensional (2D) van der Waals materials which have strong inplane covalent bonds and weak interlayer interactions such as graphene and transition-metal dichalcogenides (TMDCs) have been studied extensively because of their novel features. Two-dimensional materials can obtain atomically thin thickness film structures without dangling bonds, and this two-dimensional nature realizes advantages such as high integration, high transparency and high flexibility. Also, the quantum confinement effect and reduced screening effect arising from this two-dimensional nature often manifest themselves in features clearly different from those of their bulk counterparts. Unlike graphene, which has no electric bandgap, TMDCs have various properties such as semi-metallic, semiconducting or superconducting properties depending on the combination of transition metal and chalcogens.

Among TMDC materials, WSe₂ has received considerable attention due to its unique electronic and optoelectronic properties, such as the existence of energy band gaps in the visible and near-infrared spectral ranges, and weak or non-existent Fermi level pinning at metal-WSe₂ interfaces. WSe₂ is a semiconducting material with a tuneable bandgap depending on thickness; 1.64 eV direct bandgap in monolayer and 1.2 eV indirect bandgap in bulk. In addition, the electric carrier type of WSe₂ is controllable through thickness; p-type with thin, ambipolar with intermediated, and n-type with thick thickness.

1.3. Organic-inorganic hybrid halide perovskite

1.3.1 Brief introduction of organic-inorganic hybrid halide perovskite and methylammonium lead iodide perovskite

A hybrid organic-inorganic halide perovskites (OIHPs) has a perovskite structure denoted with the structural formula ABX₃, where A is an organic cation, B is an inorganic cation and X is a halide. OIHP has received tremendous attention in recent years due to its superior properties that depend on the combination of components for optoelectronics and electronic devices. Also, OIHPs have potential in space energy application owing to their attractive properties, such as high powerconversion efficiency, ultra-light weight, and large absorption coefficient.

1.3.2 Synthesis of MAPbI₃ perovskite

There are various ways to synthesize OIHP such as spin-coating methods, thermal evaporation methods, and mechanochemical synthesis. Out of various methods, Spin-coating methods are widely used in lab-scale fabrication because it is an easily accessible methods in terms of cost and time. However, spin-coating techniques have some challenges to be addressed such as solvent induced-damages, toxicity to human body, and producing reliable and uniform films over a large area.

On the other hand, dry synthesis methods such as thermal evaporation methods, and mechanochemical synthesis have the advantage that they do not require the use of solvents, allowing materials to be synthesized without solventinduced damage, which is related to the performance of perovskite-based devices. In addition, the dry synthesis method has several additional advantages. Evaporation methods can produce a uniform film over a large area on the wafer scale. In the case of mechanochemical synthesis, not only stoichiometric fine control is possible with very small amount of by-product, so perovskite with small defects can be synthesized. but also easy to accessible and rapid synthesis. Also, the dry synthesis method can produce OIHP films with a wide range of compositions by combining various cations and halides.

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1.4. Outline of this thesis

This thesis mainly focuses on the high energy proton irradiation effect on promising nanomaterials such as WSe₂ of transition metal dichalcogenides and MAPbI₃ of OIHP with various dose conditions. The major reason for exploring effects of proton irradiation on nanomaterials during my doctorate course is to broaden our understanding of the interaction between irradiated protons and target materials and examine the potential of device applications in a high-radiation environment such as space.

In Chapter 2, the study of the changes in the electrical characteristics of WSe₂ field effect transistor after the various dose of proton irradiation are discussed. These changes can be explained by proton irradiation-induced traps such as "positive bulk trap" in oxide layer and "negative interface trap state" at the interface between the oxide layer and WSe₂. In Chapter 3, I discuss the high energy proton radiation hardness of methylammonium lead iodide perovskite synthesized by two kind of methods; flash-evaporation and mechanochemical synthesis. The radiation hardness of proton beam varies depending on the synthesis method of perovskite because the bonding strength in the synthesized perovskite vary depending on the synthesis method. Finally, Chapter 4 summaries this thesis and suggests directions for future research.

Chapter 2. Effect of Proton Beam Irradiation on WSe₂ Ambipolar FETs

In this chapter...

The irradiation effect of high energy proton beams on tungsten diselenide (WSe₂) ambipolar field-effect transistors was investigated. We measured the electrical characteristics of the fabricated WSe₂ FETs before and after the 10 MeV proton beam irradiation with different doses of 10^{12} , 10^{13} , 10^{14} , and 10^{15} cm⁻². For low dose conditions (10^{12} , 10^{13} , and 10^{14} cm⁻²), the threshold voltages shifted to the negative gate voltage direction, and the current in the hole and electron accumulation regimes decreased and increased, respectively. However, the trends were opposite for the high dose condition, and the current in the hole and electron accumulation regimes increased and decreased, respectively. These phenomena can be explained by the combined effect of proton irradiation-induced traps and the applied gate bias condition. Specifically, irradiation-induced positive oxide traps in SiO₂ dielectric play a role in enhancing electron accumulation-induced holes near the WSe₂/SiO₂ interface act as electron trapping sites, with enhancing

hole accumulation and reducing electron accumulation in the WSe₂ channel. This work will help improve the understanding of the effect of high energy irradiation on WSe₂-based and other ambipolar nanoelectronic devices. In addition, this work shows the possibility of tuning the electrical properties of WSe₂-based devices.

2.1. Introduction

Two-dimensional (2D) semiconducting transition-metal dichalcogenides (TMDs) have gained significant attention due to their unique electronic and optoelectronic properties, the presence of energy band gaps from the visible and near infrared spectral range, and atomically thin film structures.¹⁻⁶ To date, tremendous efforts have been devoted to designing and synthesizing 2D TMDs and to exploring their potential applications.^{2,7-14} Therefore, understanding the fundamental mechanisms and controlling the properties of 2D TMDs is crucial for further versatile applications in integrated devices including energy systems, light emitting devices, field effect transistors (FETs) and photodetectors.^{3,5,6,10,15-19}

The material properties of 2D TMDs can be effectively tuned through various methods such as dimension reduction, heterostructure formation, intercalation, doping, strain engineering, and especially the irradiation effects of ions or electrons.^{3,13,19-31} In particular, most irradiation studies to date have been focused on single type charge carriers in TMDs devices.²⁹⁻³¹ Among 2D TMDs, tungsten diselenide (WSe₂) has recently been studied from the perspective of irradiation effects of ion- and electron-beams.³²⁻³⁵ WSe₂ has many excellent

properties, such as a tunable bandgap (1.2-1.67 eV), ambipolar transport behaviour, and weak or nonexistent Fermi level pinning at metal-WSe₂ interfaces; ^{3,25,36} these properties make 2D WSe₂ a promising candidate for potential device applications, including valley-based electronics, spin-electronics, and optoelectronics.^{23,37-39} To the best of our knowledge, there is no report on the electrical transport properties of WSe₂ FETs showing p-type and n-type conduction through irradiation of high energy proton beams.

In this work, we studied the effect of high energy proton beam irradiation on the ambipolar WSe₂ FETs. The devices were irradiated with 10 MeV energy protons under different dose conditions of 10^{12} , 10^{13} , 10^{14} and 10^{15} cm⁻². The electrical characteristics of the WSe₂ FETs such as current level and threshold voltages were measured before and after the proton beam irradiation, and the changes after the irradiation were analysed statistically. Additionally, we measured physicochemical properties such as lattice vibration, binding energy, and thickness of the WSe₂ flakes using Raman spectra, AFM, and X-ray photoelectron spectra. After the proton beam irradiation, the electrical properties were changed due to the irradiation-induced traps from the substrate, not from the WSe₂ itself.

2.2. Experiments

2.2.1 Device fabrication process of WSe₂ FETs

Figure 2.1 illustrates the device fabrication processes for WSe₂ ambipolar FETs. First, a highly doped p-type Si wafer with 270 nm thick SiO₂ layer was

prepared. The WSe₂ flakes were transferred onto the silicon substrate by the mechanical exfoliation using a scotch tape from a bulk WSe₂ crystal (purchased from HQ Graphene). Using an optical microscope, candidate WSe₂ flakes with a few layers thickness range from 4 to 7 nm were selected to make ambipolar type WSe₂ FETs. (Table 2.1) The thicknesses of WSe₂ flakes were measured using an atomic force microscope (AFM) (Park Systems, NX10). Then, we spin-coated a bilayer electron beam resist; first methyl methacrylate (MMA) (8.5) MAA (9% concentration in ethyl lactate) was spin-coated on the samples at 4,000 rpm for 50 s, and then the samples were baked on a hotplate at 180 °C for 90 s. Next, poly methyl methacryllate (PMMA) 950K (5% concentration in anisole) was spincoated on MMA-coated samples at 4,000 rpm for 50 s, followed by a bake on the hotplate at 180 °C for 90 s. Then, we patterned the source and drain electrodes using an electron beam lithography (JEOL, JSM-6510) and performed development process with methyl isobutyl ketone: isopropyl alcohol (MIBK:IPA) (1:3) solution for 50 s. Finally, we deposited Au (40 nm)/Ti (5 nm) as the source and drain electrodes using an electron beam evaporator.



Figure 2.1 Schematics of WSe₂ device fabrication. Adapted from Shin et al. ⁵²

2.2.2 Electrical and optical characteristics measurements

The electrical characteristics of the fabricated devices were measured in a probe station (ST-500, JANIS) using a semi-conductor parameter analyzer (Keithley 4200 SCS) in vacuum (~10⁻⁴ Torr). Raman spectra of WSe₂ flakes were measured by using a Raman spectroscopy (XperRam 200, Nanobase, Inc.) with a 532 nm laser as the excitation source. The laser power was 40 μ W with a diffraction-limited laser spot size of ~1 μ m. The XPS spectra were measured by electron spectroscopy for chemical analysis (AXIS SUPRA) in National Center for Inter-university Research Facilities.

2.2.3 Proton beam irradiation experiment

The fabricated devices were exposed to proton beams with different dose conditions of 10^{12} , 10^{13} , 10^{14} , and 10^{15} cm⁻². The proton beam facility used for this research was the MC-50 cyclotron at the Korea Institute of Radiological and Medical Sciences. The proton beam had an energy of 10 MeV with an average current 10 nA, corresponding to irradiated 6.25×10^{10} protons per seconds. All the measurement for the proton-irradiated devices was taken one day after the proton irradiation since the radioactivity had to fall below the safe level.

2.3. Results and discussion

2.3.1 Electrical characteristics

The WSe₂ FETs were fabricated with mechanically exfoliated WSe₂ flakes on heavily p-doped Si substrates with a 270 nm-thick SiO₂ dielectric layer. Fig. 2.2 (a) is the optical image of a transferred WSe₂ flake on the Si/SiO₂ substrate. The WSe₂ flake shown in Fig. 2.2 (a) was found to be ~4 nm-thick using an atomic force microscope (AFM) (Fig. 2.2 (b)). Fig. 2.2 (c) shows the optical image of a WSe₂-based device made with the flake shown in Figs. 2.2 (a) and 2.2 (b).



Figure 2.2 (a) An optical image of a multilayer WSe₂ flake. (b) An AFM image of the selected region in (a) with cross-sectional profile across a line in the image. (c) An optical image of the fabricated FET using this WSe₂ flake. (d) The transfer curves of the fabricated WSe₂ FET measured at $V_{DS} = 0.1$ V, 0.5 V, and 1 V. The output curves for (e) negative gate voltages and (f) positive gate voltages measured at different gate voltages. Adapted from Shin *et al.* ⁵²

The electrical properties of the fabricated WSe₂ FET were examined, as shown in Figs. 2.2 (d)-2.2 (f). Transfer characteristics (drain-source current versus gate voltage, $I_{DS}-V_{GS}$) (Fig. 2.2 (d)) and output characteristics (drain-source current versus drain-source voltage, $I_{DS}-V_{DS}$) (Figs. 2.2 (e) and 2.2 (f)) show a sharp increase of I_{DS} at both negative (the hole accumulation regime, marked by blue colour) and positive (the electron accumulation regime, marked by red colour) gate voltages, indicating the gate-induced charge injection from metal/WSe₂ contacts. According to Podzorov *et al.*⁴⁰, the Schottky-type FETs exhibit ambipolar characteristics by gate-controlled tunneling of injected charges (hole and electron) through the Schottky barriers formed at the metal/WSe₂ semiconductor contacts. The electrical properties of other ambipolar WSe₂ FETs show in the Fig. 2.3 and Table 2.1.



Figure 2.3 The AFM images of WSe₂ flakes and corresponding transfer curves of the WSe₂ FETs with thickness of (a) 5.1 nm, (b) 5.8 nm, and (c) 7 nm. Adapted from Shin *et al.* 52

Table 2.1 The previous studies of thickness-dependent electrical conduction typeof WSe_2 FETs

	Thickness (nm)	Electrode	Reference	
p-type	ambipolar	n-type	—	
<4	~6	>15	Ni/Au	[23]
	7–26		Ti/Au	[32]
	6		S: Ni, D: Pd	[36]
<3	~4	>5	Cr/Au	[53]
	4-7		Ti/Au	[52]

2.3.2 Proton beam dose-dependence

After electrical characterization of the fabricated WSe₂ FETs, the devices were subsequently irradiated by proton beams of 10 MeV for 10, 100, 1000, and 10000 seconds, corresponding to dose values of ~ 10^{12} , 10^{13} , 10^{14} , and 10^{15} cm⁻², respectively. The electrical characteristics of WSe₂ FETs were measured and compared before and after proton beam irradiation, as shown in <u>Fig. 2.4 and Fig. 2.5</u>.

In the transfer curves of the device measured at fixed $V_{DS} = 1$ V, both the threshold voltages for hole and electron accumulation regimes shifted to the negative gate voltage direction after the proton beam irradiation for the dose of ~10¹², 10¹³, and 10¹⁴ cm⁻² (Figs. 2.4 (a)-2.4 (c)). The drain-source current level of the hole accumulation regime decreased, whereas that of the electron accumulation regime increased. For example, after the proton beam irradiation under a 10¹² cm⁻² dose, the drain-source current decreased from 0.11 µA to 0.04 µA at $V_{DS} = 1$ V and $V_{GS} = -60$ V for the hole accumulation regime, and it increased from 0.06 µA to 0.31 µA at $V_{DS} = 1$ V and $V_{GS} = 60$ V for the electron accumulation regime (Fig. 2.4 (a)). In the low irradiation dose conditions of ~10¹², 10¹³, and 10¹⁴ cm⁻², the shifted amount of the threshold voltage seemed to increase as the dose increased (marked by blue and red arrows in Figs. 2.4 (a)-2.4 (c)).



Figure 2.4 Transfer curves of the WSe2 FETs before and after proton beam irradiation under the doses of (a) 10^{12} , (b) 10^{13} , (c) 10^{14} , and (d) 10^{15} cm⁻². Adapted from Shin et al. ⁵²



Figure 2.5 The output characteristics with proton beam irradiation dose conditions of 10^{12} , 10^{13} , 10^{14} , and 10^{15} cm⁻² for (a) hole and (b) electron accumulation regimes. Adapted from Shin *et al.* ⁵²

Correspondingly, the amount of current change in the hole and electron accumulation regimes became larger as the dose increased (Figs. 2.4 (a)-2.4 (c)). For example, after the proton beam irradiation under a 10^{14} cm⁻² dose, the drain-source current decreased from 140 nA to 4 nA at $V_{DS} = 1$ V and $V_{GS} = -60$ V for the hole accumulation regime, and it increased from 0.4 nA to 16 nA at $V_{DS} = 1$ V and $V_{GS} = 60$ V for the electron accumulation regime (Fig. 2.4 (c)). However, the WSe₂ FETs that were irradiated with a proton beam dose of 10^{15} cm⁻² showed different behaviour compared to the devices irradiated under low dose conditions (Fig. 2.4 (d)). After proton beam irradiation under a 10^{15} cm⁻² dose, the threshold voltages shifted to the positive gate voltage direction for both hole and electron accumulation regimes, and the drain-source current increased in the hole accumulation regime while it decreased in the electron accumulation regime. These dose-dependent phenomena of proton beam irradiation will be explained in detail later in terms of the combined effects by irradiation-induced charges and gate-bias condition.

2.3.3 Statistical analysis of electrical properties

To clarify these different trends between the two cases of low and high proton beam doses, we statistically investigated the changes of current and threshold voltage as a function of proton beam dose. We measured a total of 20 devices, with 4–6 devices for each proton beam irradiation condition. Figs. 2.6 (a) and 2.6 (b) summarize the changes of threshold voltages for the WSe₂ FETs after the proton beam irradiation under dose conditions of 10^{12} , 10^{13} , 10^{14} , and 10^{15} cm⁻²



Figure 2.6 Threshold voltage shift measured at $V_{DS} = 1$ V as a function of proton irradiation dose condition for (a) hole accumulation regime and (b) electron accumulation regime. The normalized I_{DS} (current ratio before and after irradiation, I_{after}/I_{before}) measured at $V_{DS} = 1$ V as a function of proton irradiation dose condition for (c) hole accumulation regime and (d) electron accumulation regime. Adapted from Shin *et al.* ⁵²

for hole and electron accumulation regimes at $V_{DS} = 1$ V, respectively. Note that the threshold voltage was determined by the value of x-intercept of the tangent line from the point that has the maximum differential value in the transfer curve. In Figs. 2.6 (a) and 2.5 (b), similar to Fig. 2.4, the threshold voltages for both hole and electron accumulation regimes moved to negative gate voltage direction for low dose conditions of ~ 10^{12} , 10^{13} , and 10^{14} cm⁻². In addition, as the irradiation dose increased, the threshold voltages shifted more. However, the threshold voltage of the devices irradiated with high dose proton beam (10^{15} cm⁻²) shifted to the positive gate voltage direction. Figs. 2.6 (c) and 2.6 (d) summarize the changes of the device current after the proton beam irradiation for hole and electron accumulation regimes, respectively.

The currents in these Figs. are the values measured at $V_{DS} = 1$ V with $V_{GS} = -60$ V for the hole accumulation regime and $V_{GS} = 60$ V for the electron accumulation regime. The device current for the hole accumulation regime (Fig. 2.6 (c)) decreased as the irradiation dose increased to 10^{14} cm⁻², and then it increased at the dose of 10^{15} cm⁻². The current showed an opposite tendency for the electron accumulation regime (Fig. 2.6 (d)); the current increased as the irradiation dose increased as the irradiation dose increased as the irradiation regime (Fig. 2.6 (d)); the current increased as the irradiation dose increased to 10^{14} cm⁻², and then it decreased at the dose of 10^{15} cm⁻². As we mentioned above, these results can be explained by the irradiation-induced positive oxide traps in the gate dielectric SiO₂ layer and electron trapping at the interface between SiO₂ and WSe₂. Note that mobility and subthreshold swing values are also summarized as a function of proton dose in Figs. 2.7 and 2.8. The mobility of the ambipolar WSe₂ FET devices was calculated using the following equation:
$$\mu = \frac{L}{W} \frac{1}{C_i V_{DS}} \frac{dI_{DS}}{dV_{GS}}$$

where *L* and *W* are the channel length and width of the FET, respectively. V_{DS} is the source–drain voltage. I_{DS} is the current flowing from source to drain, and V_{GS} is the gate voltage. C_i is the gate capacitance per unit area. But mobility and subthreshold swing values did not show a clear trend with the dose conditions as the threshold voltages or currents did.



Figure 2.7 Mobility of the devices before and after the proton beam irradiation for (a) hole and (b) electron accumulation regimes. Adapted from Shin *et al.* ⁵²



Figure 2.8 Subthreshold swing value of the devices before and after the proton beam irradiation for (a) hole and (b) electron accumulation regimes. Adapted from Shin *et al.* 5^2

2.3.4 Physicochemical properties of WSe₂ flakes

To verify whether such variations of electrical characteristics are caused by proton beam-irradiated WSe₂ flakes themselves, we examined the structural characteristics, elemental composition, and electronic state of WSe₂ flakes before and after the proton beam irradiation by Raman spectroscopy and X-ray photoemission spectroscopy (XPS). Fig. 2.9 shows the measured results for the high dose condition of 10^{15} cm⁻². Raman mapping data of the WSe₂ flake (Figs. 2.9 (a) and 2.9 (b)) show no noticeable differences before and after proton beam irradiation. Specifically, in Raman spectra (Fig. 2.9 (c)), the position and intensity of three prominent Raman peaks corresponding to lattice vibration mode of E^{1}_{2g} (254 cm⁻¹), A_{1g} (254 cm⁻¹ and 264 cm⁻¹), and B¹_{2g} (310 cm⁻¹)^{38,41,42} were not noticeably changed after proton beam irradiation. The XPS result (Fig. 2.9 (d)) shows three dominant binding energy peaks of tungsten and two dominant peaks of selenium.^{42,43} These five peaks did not show considerable change after proton beam irradiation. Note that a little redshift (by ~0.1 eV) was observed after proton beam irradiation (Fig. 2.9 (d)), which may have originated from a dipole effect due to the irradiation-induced electron trapping at the interface between SiO₂ and WSe₂ with van der Waals gap.⁴⁴⁻⁴⁶ As a result, the physicochemical characteristics of WSe₂ were not damaged by proton beam irradiation, and therefore, the change of the electrical characteristics after proton beam irradiation is not considered to be caused by the WSe₂ itself.



Figure 2.9 (a, b) Raman mapping, (c) Raman spectra, and (d) XPS spectra of WSe₂ flakes before and after proton irradiation at a dose of 10^{15} cm⁻². Adapted from Shin *et al.* ⁵²

2.3.5 Mechanism of proton irradiation effect

The experimental results can be explained by the combined effects of the proton beam irradiation-induced traps and applied gate bias condition. When highenergy protons are irradiated on the WSe₂ FETs, protons generate various kinds of traps. The high energy protons slow down mainly by an electronic stopping process, generating electron-hole pairs. After the protons sufficiently slow down, they lose the majority of the energy by a nuclear stopping process, and then they stop near the stopping depth and damage on the lattice around the stopping depth.⁴⁷⁻⁴⁹ To calculate the stopping depth, we simulated energy-loss profiles using Stopping and Range of Ions in Matter (SRIM) software. According to the SRIM results, most of the protons with 10 nA current and 10 MeV energy would stop near 700 µm from the surface (Fig. 2.10). Since the thickness of the device is approximately 500 µm (4–7 nm for the WSe₂ flake, 270 nm for the SiO₂ layer, and 500 µm for the Si layer), most of the protons would penetrate through the entire device structure and electron-hole pairs are generated along the paths.



Figure 2.10 The energy loss profiles of the protons simulated by SRIM. The inset image is a zoomed result for the range of 0-400 nm from the top surface which includes the oxide layer. Adapted from Shin *et al.* 5^{2}

The SiO₂ layer is the most sensitive to the ionizing process. After electron-hole pairs are generated, they recombine for a few picoseconds, but some of the holes remained near the generated location while electrons are swept away much faster.^{50,51} The trapped holes in the SiO₂ layer (denoted as "positive oxide traps") cause the modulation of gate electric field, resulting in the depletion effect of holes and enhancement effect of electrons in the WSe₂ channel (Figs. 2.11 (a) and 2.11 (c)). Some of the generated holes undergo an anomalous stochastic hopping transport through the SiO₂ layer and reach the WSe₂/SiO₂ interface under the high

dose condition (Figs. 2.11 (e) and 2.11 (g)). These holes that reach the interface can cause the formation of a layer of electron trapping sites, resulting in the enhancement effect of holes and depletion effect of electrons in the WSe₂ channel (Figs. 2.11 (e) and 2.11 (g)). As a result, positive oxide traps play a role in enhancing electron accumulation and reducing hole accumulation in the WSe₂ channel, whereas the electron trapping sites near the WSe₂/SiO₂ interface (denoted as "interface electron trap sites") enhance hole accumulation in the WSe₂ channel. Because the interface electron trap sites require a longer proton irradiation time than the positive oxide traps, the effect of the positive oxide traps in SiO_2 bulk is more influential for low dose conditions $(10^{12}, 10^{13}, \text{ and } 10^{14} \text{ cm}^{-2})$ (Figs. 2.11 (a)– 2.11 (d)). In the energy band diagrams, due to the positive oxide traps, electrons are efficiently injected into the WSe₂ channel under positively applied gate voltage (Fig. 2.11 (b)) while it becomes difficult to inject holes into the WSe_2 channel under negatively applied gate voltage (Fig. 2.11 (d)). Thus, the threshold voltage shifts to the negative gate voltage direction (Figs. 2.4 (a)-2.4 (c), 2.6 (a) and 2.6 (b)). As the proton beam irradiation dose increases, more positive oxide traps are generated. Therefore, the threshold voltage shift can be larger because the positive oxide traps increase as the proton beam irradiation dose increases, as shown in Figs. 2.6 (a)-2.6 (b). However, in the high dose condition of 10^{15} cm⁻², the effect of interface electron trap sites becomes dominant, and therefore, the threshold voltage shifts to the positive gate voltage direction (Figs. 2.4 (d), 2.6 (a) and 2.6 (b)). Here, it becomes difficult to inject electrons into the WSe_2 channel (Fig. 2.11 (f)) and easier to inject holes into the WSe₂ channel (Fig. 2.11 (h)).



Figure 2.11 Schematic illustrations of device structure and energy band diagrams depicting the conduction mechanism after proton beam irradiation. For low dose conditions $(10^{12}, 10^{13}, \text{ and } 10^{14} \text{ cm}^{-2})$, the effect of positive oxide traps is dominant, resulting in the enhancement of electron accumulation in the channel (a, b, c, d). For the high dose condition $(10^{15} \text{ cm}^{-2})$, the effect of interface electron trap sites is dominant, resulting in the enhancement of hole accumulation in the channel (e, f, g, h). Adapted from Shin *et al.* ⁵²

2.4 Conclusions

In summary, we investigated the irradiation effect of a high-energy (10 MeV) proton beam on the WSe₂ FETs with ambipolar characteristics. The electrical properties such as the current and threshold voltages of the WSe₂ devices were changed after proton beam irradiation, depending on the beam dose conditions due to the combing effect of the proton beam irradiation-

induced charges and applied gate bias. Under low dose conditions $(10^{12},$ 10^{13} , and 10^{14} cm⁻²), the positive trapped holes in the SiO₂ dielectric layer produces the modulation of effective gate field on electron and hole accumulation to the WSe₂ channel, resulting in a threshold voltage shift towards the negative gate voltage direction. Under the high dose condition $(10^{15} \text{ cm}^{-2})$, the irradiation-induced holes near the WSe₂/SiO₂ interface act as electron-trapped sites and enhance hole accumulation in the WSe₂ channel, resulting in a threshold voltage shift towards the positive gate voltage direction. This study may provide a method to control the electrical properties of WSe₂ FETs through proton beam irradiation. Furthermore, this study may contribute to a deeper understanding of the influence of highenergy proton beams on ambipolar WSe₂-based nanoelectronic devices and be helpful for application of TMD-based devices in harsh radiation environments such as space.

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Chapter 3. Proton Irradiation Effects on Mechanochemically Synthesized and Flashevaporated Hybrid Organic-inorganic Lead Halide Perovskites

In this chapter...

A hybrid organic-inorganic halide perovskite is a promising material for developing efficient solar cell devices, with potential applications in space science. In this study, we synthesized methylammonium lead iodide (MAPbI₃) perovskites via two methods: mechanochemical synthesis and flash evaporation. We irradiated these perovskites with highly energetic 10-MeV proton-beam doses of 10^{11} , 10^{12} , 10^{13} , and 4×10^{13} protons/cm² and examined the proton irradiation effects on the physical properties of MAPbI₃ perovskites. The physical properties of the mechanochemically synthesized MAPbI₃ perovskites were not considerably affected after proton irradiation. However, the flash-evaporated MAPbI₃ perovskites showed a new peak in X-ray diffraction and an increased fluorescence lifetime in time-resolved photoluminescence under high-dose conditions, indicating considerable changes in their physical properties. This difference in behavior between MAPbI₃ perovskites synthesized via the abovementioned two methods may be attributed to differences in radiation hardness associated with the bonding strength of the constituents, particularly Pb–I bonds. Our study will help to understand the radiation effect of proton beams on organometallic halide perovskite materials.

3.1. Introduction

Organic-inorganic hybrid halide perovskites (OIHPs) have promising applications in optoelectronic and electronic devices, including solar cells, lightemitting diodes, lasers, photodetectors, and transistor [1-12], owing to their attractive properties, such as high power-conversion efficiency, large charge-carrier diffusion length, large absorption coefficient, and tunable direct bandgap [13,14]. However, the stability of hybrid perovskites needs to be investigated under ambient and harsh environmental conditions to demonstrate their practical applicability. In particular, given the prospect of high-efficiency perovskite solar cells being used in space environments, some researchers have studied the radiation tolerance of perovskite materials and devices to high-energy particles, such as cosmic rays, under space environments. For example, Lang et al. investigated the radiation hardness of inverted methylammonium lead iodide (MAPbI₃) perovskite solar cells and found that these devices could tolerate 68-MeV proton-beam doses up to 10¹² protons/cm² [15]. Miyazawa et al. found that lead halide perovskite solar cells could tolerate proton irradiation dose up to 10¹⁵ particles/cm² with 50-keV proton beam which enable to implant proton particles within the perovskite layer causing lattice deformation [16]. However, the effect of isolating perovskites from the influences of other layers of the devices or interfaces that may affect the radiation hardness has not been adequately reported. Lang et al. irradiated lead halide perovskite/copper indium gallium selenide (CIGS) and perovskite/silicon tandem solar cells and attributed most of the device efficiency losses to the CIGS and silicon layers and not the perovskite layer [17]. Therefore, exclusive research to investigate the apparent radiation hardness of the perovskite layer itself is necessary to avoid the influence of other surrounding layers.

In this study, we synthesized organic-inorganic hybrid MAPbI₃ perovskites via two methods: mechanochemical synthesis and flash evaporation. Then, we investigated the effects of irradiation on the physical properties of MAPbI₃ perovskites under various high-energy proton-beam doses. For this purpose, the physical properties of perovskites synthesized via two methods were characterized systematically before and after 10-MeV proton beam irradiation doses of 10^{11} , 10^{12} , 10^{13} , and 4×10^{13} protons/cm² using X-ray diffraction (XRD), ultraviolet-visible (UV-Vis) absorbance spectroscopy, and photoluminescence. Mechanochemically synthesized MAPbI₃ perovskites did not exhibit considerable changes in their physical properties after the proton irradiations, indicating that this perovskite can tolerate dose conditions in this study. Conversely, flash-evaporated MAPbI₃ perovskites exhibited considerable changes in their physical properties after the proton irradiations, indicating that this perovskites exhibited considerable changes in their physical properties after the proton irradiations, indicating that this perovskite can tolerate dose conditions in this study. Conversely, flash-evaporated MAPbI₃ perovskites exhibited considerable changes in their physical properties after the proton irradiation our understanding of the radiation tolerance of perovskite under high-energy proton-beam irradiation to estimate the potential of perovskites in space applications.

3.2. Experiments

3.2.1 Mechanochemical synthesis of MAPbI₃ perovskite

MAPbI₃ perovskite powder was synthesized mechanochemically, as illustrated in figure 3.1. Figure 3.1 (a) shows an optical image of the precursors 1.59-g methylammonium iodide (MAI) (white powder) and 4.61-g lead iodide (PbI₂) (yellow powder), ground using stainless steel balls with diameters of 1.27, 0.95, and 0.63 cm (4, 4, and 60 balls, respectively) in an alumina jar. The molecular structures of MAI and PbI₂ are schematically illustrated in figure 3.1 (d). The alumina jar containing the mixture was ball milled at 300 Hz for 5 h (BD4530, LK Lab) (Figure 3.2).



Figure 3.1 (a) Optical image of methylammonium iodide (MAI) and lead iodide (PbI₂) precursor powders in alumina jar. (b) Schematic of mechanochemical synthesis of MAPbI₃ perovskite through a ball mill. (c) Optical image of mechanochemically synthesized MAPbI₃ perovskite powder. (d) Structures of MAI and PbI₂ precursors. (e) Mechanism of mechanochemical reaction from reactants to products via soft ball milling. (f) Structure of MAPbI₃ perovskite unit cell. Adapted from Shin *et al.* ⁵²



Figure 3.2 (a) Image of ball mill system. (b) Synthesized perovskite powder.

Adapted from Shin et al. 52

3.2.2 Synthesis of MAPbI₃ perovskite films via flash evaporation

As, previously reported, MAPbI₃ perovskite films were synthesized via a flash evaporation method [18, 19]. First, 500-µm-thick glass substrates were cleaned via sonication in acetone, isopropyl alcohol, and deionized water for 10 min each. Then, the substrates were treated with O₂ plasma (30 sccm, 50 W) for 120 s using a reactive ion etcher (RIE; AFS-R4T, AllForSystem). These treated substrates were transferred to a vacuum chamber and placed in a substrate holder 30 cm above a source boat for flash evaporation (**see figure 4(a)**). To deposit MAPbI₃ perovskite film, we applied a high current of 100 A to rapidly heat the 650-mg mechanochemically synthesized MAPbI₃ powder and 167-mg MAI powder placed in the source boat to evaporate the source powder mixture.

3.2.3 Characterization of mechanochemically synthesized perovskite powder and flash evaporated perovskite film

The optical images and energy-dispersive spectra of the synthesized perovskite powder and film samples were acquired via field-emission scanning electron microscopy (FESEM; JSM-7800F Prime, JEOL Ltd). Powder or film XRD spectra were measured using an X-ray diffractometer (SmartLab, Rigaku) in the National Center for Interuniversity Research Facilities of Korea. Photoluminescence (PL) spectra and time-resolved (Tr) PL spectra were measured using the XperRAM 200 with a laser of 532-nm excitation wavelength and XperRF (Nanobase Inc.) with a pulsed laser that has a 405-nm excitation wavelength, 5000-kHz repetition rate, and 1- μ W power, respectively. The absorbance spectra were acquired via UV–Vis spectroscopy (V-770, Jasco).

3.2.4 Proton irradiation

We used the MC-50 cyclotron at the Korea Institute of Radiological and Medical Sciences for performing proton beam irradiation experiments. The 10-MeV proton beam irradiation source generated by 2×10^{10} protons per second per square centimeter, corresponding to a 10-nA average current had a ~1-cm spot size. Proton irradiation was performed in a vacuum environment. In order to irradiate the proton on the powder samples, the mechanochemically synthesized perovskite powder was placed on 15 µm thick Al foil. 140 mg perovskite powder was put on the Al foil with an area of 1.4 cm × 1.4 cm with a thickness of ~150 µm. The flashevaporated perovskite film was directly attached to the equipment without any additional packing steps. The perovskite powder and films were irradiated at 10^{11} , 10^{12} , 10^{13} , and 4×10^{13} protons/cm² proton-beam doses, which correspond to irradiation times of 5, 50, 500, and 2000 s, respectively. Cosmic ray proton flux varies depending on distance from earth, and the dose of 4×10^{13} protons/cm² equivalent to the number of protons encountered in space over 10 years [20, 21]. To allow the radioactivity to fall to a safe level, the proton-irradiated samples were retained in the facility for one day after the completion of proton beam irradiation experiments.

3.3. Results and discussion

3.3.1 Characterization of Mechanochemically synthesized perovskite

We used an electrically rotating ball mill in a so-called "planetary instrument," a simplified version of which is shown in figure 3.1 (b), among various mechanochemical synthesis methods, such as hand-grinding or automated mixer mill methods. This method facilitates adequate controllability of rotation frequency and grinding force [22]. As the alumina jar rotates, mechanical energy, such as compression, shear, or friction, formed between the stainless balls or between the balls and the jar is transferred to the mixture (figure 3.1 (b) and 3.1 (e)).



Figure 3.3 (a, b) SEM images of MCS MAPbI₃ perovskite powder. (c) XRD spectrum of MCS MAPbI₃ perovskite powder. (d) Steady-state photoluminescence (PL) spectrum of MCS MAPbI₃ perovskite powder. Adapted from Shin *et al.* ⁵²

When sufficient energy is transferred, the chemical reaction occurs, and high-purity perovskite samples can be obtained as a form of dry colloidal powder. This method does not require dissolving reagents, and fresh-reactant surfaces are constantly exposed during the synthesis process [22-24]. Figure 3.1 (c) presents an optical image that shows that the mechanochemically synthesized (denoted as "MCS") MAPbI₃ perovskite powder is black as is known. Figure 3.1 (f) shows the crystal structure of the MCS perovskite, which is the tetragonal phase. Note that

flash-evaporated perovskite has the same tetragonal structure [25-27].



Figure 3.4 (a) SEM image and (b-f) EDS mapping images of MAPbI₃ perovskite powder sample. (b) Combination map of Pb, I, C, and N and map of (c) Pb, (d) I,

(e) C, and (f) N. (g) EDS spectral line pattern of MAPbI₃ powder sample. Adapted from Shin *et al.* ⁵²

To verify whether MCS MAPbI₃ perovskite powder was well synthesized, we characterized its elemental composition and structural properties as well as photophysical properties using SEM, energy-dispersive X-ray spectroscopy (EDS), powder XRD, and PL. Figures 3.3 (a) and 3.3 (b) show the SEM images of the MCS MAPbI₃ perovskite powder with a particle size of tens to hundreds of nanometers. Uniform spatial distributions of elements such as C, Pb, and I were confirmed via EDS results (supplementary figure 3.4). The powder XRD peaks indicated that the MCS perovskite powder have a lattice structure of MAPbI₃ perovskite phase without remaining precursor PbI₂ that did not participate in the reaction (figure 3.3 (c)) [28, 29]. Figure 3.3 (d) shows a PL peak at ~770 nm from the steady-state PL emission with an excitation source of 532 nm, which corresponds to the known ~1.6-eV bandgap value of the MCS MAPbI₃ perovskite powder [28, 30]. Based on these experimental results, we confirmed that the MCS MAPbI₃ perovskite powder was successfully synthesized.

3.3.2 Proton irradiation hardness of MCS perovskite

Then, the MCS MAPbI₃ perovskite powder was irradiated by 10-MeV proton beams under irradiation doses of 10^{11} , 10^{12} , 10^{13} , and 4×10^{13} protons/cm²,



respectively. We compared the physical properties of MCS MAPbI₃ perovskite

Figure 3.5 (a) XRD, (b) UV-Vis absorbance, (c) PL, and (d) TRPL spectra of MCS MAPbI₃ perovskite powder samples before and after proton irradiation at doses of 10^{11} , 10^{12} , 10^{13} , and 4×10^{13} protons/cm². Adapted from Shin *et al.* ⁵²

powder systematically before and after proton irradiation to investigate the effects of proton irradiation. Figure 3.5 (a) shows the XRD spectra of the MCS MAPbI₃ perovskite powder before proton irradiation (denoted as "pristine") and after various proton irradiation doses. The XRD spectra did not indicate any considerable changes before and after proton irradiation, which suggests that the proton beam irradiation did not considerably affect the MCS MAPbI₃ perovskite powder. Furthermore, no conspicuous changes beyond the measurement error range were observed in the UV-Vis absorbance (figure 3.5 (b)) and PL results (figure 3.5 (c)) of the MCS MAPbI₃ perovskite powder before and after proton irradiation although PL spectra seem to have slight variation in the peak width. Note that the PL spectra were the average values of point-scanned PL at four to five different positions and normalized because the intensity varies according to the location of the powder sample.

The MCS MAPbI₃ perovskite powder was also characterized by TRPL, which has proven to be a reliable and suitable tool for extracting minority-carrier lifetimes [31-33]. Figure 3.5 (d) shows TRPL decay curves with excitation pulse laser of 405 nm wavelength and 5000 kHz repetition rate for the MCS MAPbI₃ perovskite powder. We analyzed these curves according to the following equation [34, 35]:

$$I(t) = a_1 e^{-t/\tau_1} + a_2 e^{-t/\tau_2} + b, \qquad (1)$$

where a and τ are the amplitude and fluorescence lifetimes of the TRPL decay curve, respectively, and b is a fitting parameter. The subscripts 1 and 2 represent early- and late-stage recombination processes, respectively. The detailed fitting results are provided in figures 3.6, 3.7 (a) and table 3.1.

Particularly, we found that both the τ_1 and τ_2 decreased with increasing irradiation dose. For the pristine case, τ_1 was found to be ~3.53 ns, which

decreased from ~2.78 to ~2.15 ns for dose from 10^{11} to 4×10^{13} protons/cm². Similarly, τ_2 was found to be ~28.4 ns for the pristine case, which decreased from ~25.0 to ~7.82 ns for dose from 10^{11} to 4×10^{13} protons/cm². In other words, higher



Figure 3.6 TRPL spectra and fitting curves of MAPbI₃ perovskite powder samples of (a) before irradiation and (b-e) after proton irradiation under dose conditions of (b) 10^{11} , (c) 10^{12} , (d) 10^{13} , and (e) 4×10^{13} protons/cm². The fitting parameters of τ_1 and τ_2 are in the unit of ns. Adapted from Shin *et al.* ⁵²

dose irradiation of the samples more accelerated the recombination process (i.e., τ values decreased). Overall, the results presented in figure 3 indicate that the proton irradiation did not cause considerable structural changes in the MCS MAPbI₃ perovskite powder (figures 3.5 (a)-3.5 (c)); however, it accelerated the recombination process. This finding suggests that the proton beams might generate



Figure 3.7 Fluorescence lifetimes (τ_1, τ_2) as a function of proton beam dose for (a) MCS and (b) flash evaporated perovskites. Adapted from Shin *et al.* ⁵²

 Table 3.1 The fluorescence lifetime of TRPL fitted to Equation (1) in the original manuscript. Adapted from Shin *et al.* 52

Dose (protons/cm ²)	MCS		Flash evaporation	
	τ_1 (ns)	$\tau_2(ns)$	τ_1 (ns)	τ_2 (ns)
Pristine	2.80	18.98	1.91	11.23
1011	2.28	17.19	1.04	4.59
10 ¹²	2.24	16.86	1.27	4.44
10 ¹³	2.52	12.43	1.53	7.63
4×10^{13}	2.02	7.27	4.71	20.44

shallow traps that can allow trap-mediated recombination (figure 3.5 (d)). A detailed discussion of this point is offered later. We also measured EDS after proton irradiation under dose of 4×10^{13} proton/cm² (figure 3.8). In our results, no noticeable change in composition was observed between before and after proton irradiation on MCS perovskite powder, considering the measurement error range of the system.



Figure 3.8 EDS results of MCS perovskite powder (a) before and (b) after irradiation under dose of 4×10^{13} proton/cm². Adapted from Shin *et al.* ⁵²

3.3.3 Proton hardness of flash-evaporated perovskite

We synthesized MAPbI₃ perovskite films via flash-evaporation and investigated the proton irradiation effect. The flash evaporation process is schematically illustrated in figure 3.9 (a). Detailed information regarding flash evaporation is provided in the Experimental section. This flash evaporation method offers the advantages of rapid, solvent-free deposition for a large-scale production of OIHP films [18, 36]. The flash-evaporated MAPbI₃ perovskite film in this study was ~150-nm-thick with a grain size of tens to hundreds of nanometers (figure 3.9 (b) and figure 3.10).



Figure 3.9 (a) Schematic of MAPbI₃ perovskite film deposition by flash evaporation. (b) SEM image of flash-evaporated MAPbI₃ perovskite film. (c) XRD, (d) UV-Vis absorbance, (e) PL, and (f) TRPL spectra of flash-evaporated MAPbI₃ perovskite film samples before and after proton irradiation. Adapted from Shin *et al.* ⁵²



Figure 3.10 Cross-sectional SEM of flash evaporated perovskite films. Adapted from Shin *et al.* ⁵²

Similarly, the physical properties of flash-evaporated MAPbI₃ perovskite films were characterized using XRD, UV-Vis absorbance, PL, and TRPL before and after proton irradiation. Figure 3.9 (c) shows the XRD spectra of flashevaporated perovskite film before and after proton irradiation. In this figure, the "pristine" and "control" curves represent the XRD spectra of the sample before proton irradiation, and that of the same pristine sample stored with other protonirradiated samples under the same conditions, respectively. Note that all the samples were stored in a desiccator whose pressure was maintained at approximately 1 torr during travel between the laboratory and proton irradiation facility. Similar to the XRD spectra for the MCS MAPbI₃ perovskite powder (figure 3.3 (c) and figure 3.5 (a)), the XRD spectrum for the pristine $MAPbI_3$ perovskite film shows the peaks corresponding to the well-synthesized MAPbI₃ perovskite lattice structure (figure 3.9 (c)), with broader XRD peak width than MCS powder, because the average grain size of flash-evaporated film (~17.12 nm) is smaller than MCS powder (~34.89 nm) which is obtained by the Scherrer equation [37]. It is noteworthy that the 2-theta value for the peaks corresponding to the flash-evaporated MAPbI₃ perovskite film decreased by $\sim 0.1^{\circ}$ compared with those of the MCS MAPbI₃ perovskite powder samples (figure 3.11).


Figure 3.11 XRD of pristine perovskites synthesized by MCS and flash evaporation. Adapted from Shin *et al.* 52

This change may suggest that the MCS MAPbI₃ powder possesses a PbI₆ framework of superior compactness [25] with a less defective nature [38, 39] than that of the flash-evaporated MAPbI₃ film. The defect in iodide, which is significantly related to the stability of the perovskite, locally increases the electrostatic repulsion of lead cations in PbI₆ framework, thereby increasing the lattice size [25]. The broad peak around 25° can be attributed to the glass substrate.

It is noteworthy that low-dose (10^{11} , 10^{12} , and 10^{13} protons/cm²) proton irradiation did not considerably affect the XRD spectra of the flash–evaporated MAPbI₃ perovskite films whereas high-dose (4 × 10^{13} protons/cm²) proton irradiation produced a new peak around 39.2° (red circled in figure 3.9 (c)), which indicates the presence of PbI_2 impurity generated via a high-dose proton irradiation in our samples[40].



Figure 3.12 Areal PL mapping of flash-evaporated perovskite films for (a) before and (b) after proton irradiation of 4×10^{13} proton/cm². PL area mapping performed using a 1 µW power laser with a wavelength of 532 nm for 500 ms per point. We scanned area of $32 \times 36 \,\mu\text{m}^2$ and $40 \times 40 \,\mu\text{m}^2$ for the samples before the irradiation and after the irradiation, respectively, with a pixel spacing of 4 µm to prevent the overlapping laser cross-section. Adapted from Shin *et al.* ⁵²

Figures 3.9 (d)-3.9 (f) show the UV-Vis absorbance, PL, and TRPL data for the flash evaporated perovskite films before and after the proton irradiation. As shown in figure 3.9 (d), the absorbance onset position did not change much for all irradiation conditions, but the slope of the spectrum below ~750 nm became smaller after the proton irradiation of high dose condition of 4×10^{13} proton/cm². Such change can also be associated with the slight deformation of MAPbI₃ which introduces additional trap sites in the bandgap, as previously reported[41]. The absorbance onset in the ~770-nm flash-evaporated MAPbI₃ perovskite films differed from that of the ~820-nm MCS MAPbI₃ perovskite powder samples (figures 3.5 (b) and figure 3.9 (d)). This disparity suggests smaller bandgap of MCS MAPbI₃ perovskite powders, which is associated with the less defects nature and superior compactness PbI₆ framework [25].

Figure 3.9 (e) shows the PL data for the flash-evaporated MAPbI₃ perovskite film before (pristine) and after proton irradiation at the highest dose (4 \times 10^{13} protons/cm²). Note that the PL spectra of the flash-evaporated film are the average values of area scan (figure 3.12). The PL peak positions did not differ but the intensity decreased by approximately 90% after high-dose (4 \times 10¹³ protons/cm²) proton irradiation compared with that of the pristine sample. This can be attributed to the generated deep traps, which act as nonradiative recombination centers because the shallow traps such as Pb vacancies and MA interstitials, which are majority traps with the lowest defect formation energy in MAPbI₃ perovskite, do not act as non-radioactive recombination centers unlike deep defects [42, 43]. Notably, unlike the normalized PL spectra of the powder samples (figure 3.5 (c)), the PL spectra of the flash-evaporated MAPbI₃ perovskite films are not normalized because the intensity does not vary according to the location of the uniform film sample. Thus, the intensity comparison is meaningful. The PL peak position for the \sim 760-nm flash-evaporated MAPbI₃ perovskite films differed slightly from that of the \sim 770-nm of the MCS MAPbI₃ perovskite powder samples (figure 3.5 (c) and figure 3.9 (e)). This disparity is consistent with the trend in absorbance data

between the two sample types (figure 3.5 (b) and figure 3.9 (d)).

Figure 3.9 (f) shows the TRPL data for the flash-evaporated MAPbI₃ perovskite films analyzed using equation (1). As the proton-irradiation dose increased, the τ

values were found to decrease relative to the pristine sample and then increase



Figure 3.13 TRPL spectra and fitting curves of MAPbI₃ perovskite films of (a) before irradiation and (b-e) after proton irradiation under dose conditions of (b)

10¹¹, (c) 10¹², (d) 10¹³, and (e) 4×10^{13} protons/cm. The fitting parameters of τ_1 and τ_2 are in the unit of ns. Adapted from Shin *et al.* ⁵²

 Table 3.2 Coefficient of variations of fluorescence lifetime. Adapted from Shin *et*

 al. 52

	MCS		Flash evaporation	
	τ_1 (ns)	τ_2 (ns)	τ_1 (ns)	τ_2 (ns)
Coefficient of variations	0.11	0.29	0.64	0.61

(figures 3.13, 3.7 (b) and table 3.1). Particularly, τ_1 was ~1.91 ns for the pristine case, which decreased to ~1.04 ns at dose of 10^{11} protons/cm² and increased from ~1.27 to ~4.71 ns at doses from 10^{12} to 4×10^{13} protons/cm², respectively. Similarly, τ_2 was ~11.23 ns for the pristine case, which decreased to ~4.59 and ~4.44 ns at doses of 10^{11} and 10^{12} protons/cm² and increased to 7.63 and 20.44 ns at doses of 10^{13} and 4×10^{13} protons/cm², respectively. These results imply that the recombination process was accelerated when the samples were irradiated at low doses because of the generation of shallow traps. Conversely, the recombination process because of the samples were irradiated at high doses because of the generation of shallow traps. Conversely, the recombination process of the generation of shallow traps. Conversely, the recombination process because of the samples were irradiated at high doses because of the generation of shallow traps. Conversely, the recombination process because of the samples were irradiated at high doses because of the generation of deep traps. Note that we obtained the coefficients of variations of fluorescence lifetime (table 3.2) to see the degree of variability. According to this,

the variation in τ_2 is similar to that of τ_1 . Also we conducted the EDS after proton irradiation for the flash-evaporated perovskite film under a dose of 4×10^{13} proton/cm² (figure 3.14). Similar to the MCS perovskite sample, a noticeable change in composition after proton irradiation was not observed.



Figure 3.14 EDS results of flash-evaporated perovskite film (a) before and (b) after irradiation under dose of 4×10^{13} proton/cm². Adapted from Shin *et al.* ⁵²

3.3.4 SRIM results and proton matter interaction

Figure 3.15 (a) shows the energy-loss depth profiles of a 10-MeV proton beam based on the simulation results performed using Stopping and Range of Ions in

Matter (SRIM) software, which is a computer program that calculates the interactions of energetic ions with target matter.



Figure 3.15 Energy loss profiles of protons simulated via SRIM (a) in the full range and (b) in the range of $0-200 \mu m$. (c) Schematics depicting the effects of proton beam irradiation on perovskite. Left image shows schematic for pristine perovskite, two images in the middle show schematics for perovskite irradiated under a low dose (left middle), in which the effect of hydrogen vacancies is dominant, and a high dose (right middle), in which the effects of lead and iodide vacancies increase. Right image shows schematic of self-healed perovskite in which hydrogen atoms released from bonds reassume their proper lattice positions. Adapted from Shin *et al.* ⁵²

There are two main processes in which a high-energy proton beam loses its energy. First, the high-energy protons are slowed down during the electronic stopping process, losing some energy and creating electron-hole pairs. Then, after the protons sufficiently slow down, they lose most of the energy during the nuclear stopping process and stop near the stopping depth, damaging the lattice of the target material [44-47]. Therefore, knowing the thickness of the target material compared with the stopping depth is necessary. According to the SRIM simulation results, as shown in figure 3.15 (a), most protons have a stopping depth near 650 μ m, which greatly exceeds the thickness of the perovskite samples (150 μ m of MCS MAPbI₃ perovskite powder samples or 150 nm of flash-evaporated MAPbI₃ perovskite film samples). Most protons would pass through the samples while transferring energy via the ionization process in the material, creating homogeneous defects throughout the entire perovskite material (figure 3.15 (b)).

When the proton beam irradiates perovskite, it can create electron-hole pairs along the irradiation path and break the bonds in the perovskite material. In this case, covalent bonds in the organic cation part (i.e., methylammonium, CH_3 – NH_3^+), such as the C–H and N–H bonds, are vulnerable to protons because they have low bonding energy (left middle image of figure 3.15 (c)). However, these broken-bond defects comprising mainly hydrogen vacancies in place of these bonds do not significantly degrade the material, because the wandering hydrogen atoms that come off the bonds have high mobility so they can take the proper lattice position, which is called as 'self-healing' process (right image of figure 3.15 (c)) [15, 48-51]. As the proton irradiation dose increases, constituent atoms of high

bonding energy (i.e., Pb or I atoms in the PbI₆ framework) detach and remain as defects rather than relax back to their lattice positions (right middle image of figure 3.15 (c)). The difference in radiation hardness between MCS MAPbI₃ perovskite powder and flash-evaporated MAPbI₃ perovskite film samples can be explained in terms of the bonding strength in the PbI₆ framework. As previously reported [25], the Pb–I bonds in MCS MAPbI₃ perovskite powder are stronger and shorter than those in flash-evaporated MAPbI₃ perovskite films. This difference is consistent with our results in terms of peak position differences obtained in XRD, PL, and UV-Vis absorbance spectra (figure 3.11 for XRD, figure 3.5 (c) vs. figure 3.9 (d) for UV-Vis absorbance).

3.4 Conclusions

We investigated the radiation hardness of organic–inorganic halide MAPbI₃ perovskites synthesized via mechanochemical synthesis and flash evaporation by measuring their physical and optical properties before and after the 10-MeV proton irradiation at various doses. The MCS MAPbI₃ perovskite powder samples did not exhibit considerable changes in their physical properties after the proton irradiation, indicating their radiation tolerance. Conversely, flash-evaporated MAPbI₃ perovskite film samples exhibited noticeable changes after irradiation at high doses. The radiation hardness difference between perovskite synthesis methods can be explained by the differences in the bonding strength of the PbI₆ frameworks in the irradiated perovskites. In particular, the Pb–I bonds in MCS MAPbI₃ perovskite powder are stronger than those in the flash-evaporated MAPbI₃ perovskite film. This study helps in expanding the scientific understanding of the radiation hardness of perovskite material and demonstrates the high radiation tolerance of MCS MAPbI₃ perovskite. Further study on an MCS MAPbI₃ perovskite powder that can be converted to a film without losing its properties will enhance the application potential of MAPbI₃ perovskite for devices in harsh radiative environments, such as space.

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Chapter 4. Summary

In this dissertation, I have dealt with the study on the effect of high energy proton beam irradiation on nanomaterials such as the WSe₂ of two-dimensional material and MAPbI₃ of metal halide perovskite.

First, I investigated the proton irradiation effect on ambipolar WSe₂ field effect transistors by analysing the electrical characteristics before and after the irradiation. Proton induced changes in electrical characteristics showed different trends according to the proton beam dose. In low dose conditions, a positive gate voltage was induced due to the "positive bulk trap" and in high dose condition, a negative gate voltage was induced due to the "negative interface trap state".

Secondly, the proton radiation hardness on organic-inorganic hybrid halide perovskite were studied. MAPbI₃ perovskite were synthesized by two kinds of methods; flash evaporation and mechanochemical synthesis. The physicochemical characteristics of the synthesized perovskite were systematically measured before and after the high-energy proton beam irradiation. Effects of the proton beam irradiation, such as radiation hardness, depended on the synthesis method. This study will help expand the understanding of the effect of high energy proton beam irradiation and suggests the possibility of application in radiation applications.

국문 초록

고에너지 양성자 조사가 나노 물질(이차원 텅스텐 다이셀레나이드와 메틸암모늄 납 할라이드

페로브스카이트)에 미치는 영향

신지원

서울대학교 물리천문학부

트랜지스터나 태양 전지와 같은 전자 소자에 대한 양성자 빔 조사의 효과에 대한 연구는 크게 두 가지의 이점을 가지고 있습니다. 첫 번째는 양성자 빔을 통하여 소자의 전기적 또는 광학적 성능을 제어할 수 있다는 점입니다. 양성자 빔의 에너지에 따라, 고에너지의 양성자 빔을 사용하여 소자에 결함을 유도하여 전류 레벨을 조절하거나 저에너지의 양성자 빔을 사용하여 계층 구조 물질을 단층으로 박리시켜 광학적 특성을 크게 향상시킬 수 있습니다. 그러나 이러한 현상에 대한 미커니즘을 명확히 하기 위해서는 추가적인 연구가 필요합니다. 두번째는 우주와 같은 고방사선 환경에서 전자 소자의 활용 가능성을

탐색할 수 있다는 점입니다. 지난 6월, 누리호가 성공적으로 발사됨으로 하국은 독자적 발사체를 우주로 쏘아 올린 7번째 나라가 되었습니다. 누리호는 순수 국산 기술로 만들어진 우주 발사체라는 것에 있어 의미가 큽니다. 미래 우주 산업은 태양광 산업, 우주 관광산업, 관측 산업 등, 여러 가능성을 품고 있어 국내외 막론하고 큰 관심을 받고 있는 분야입니다. 이를 나타내듯이 지난 50년간 평균적으로 연간 100 여개의 인공위성이 발사된 것에 반해 최근에 급격한 우주 산업의 성장으로 2021년에는 1400개 이상의 위성이 발사되었습니다. 이에 따라 우주 환경에서 활용가능한 전자 소자에 대한 수요가 증가하고 있습니다. 이 때, 우주선을 구성하는 감마선, 전자, 중성자 등 다양한 요소 중에서 양성자가 매우 높은 비율을 차지하기 때문에 양성자에 대한 영향 연구가 매우 중요합니다. 제가 연구한 물질 중 하나인 유기-무기 할로게화 납 페로브스카이트는 높은 전력 변환 효율을 가지고 있어 차세대 태양 전지 물질로 주목 받으며 많은 연구가 이루어지고 있으며 최근 실제 우주에서 시범 적용 단계에 이르렀습니다. 그러나 안정성 및 성능 면에서 최대 활용을 위하여 더 많은 연구가 필요합니다.

이러한 측면에서, 본 학위논문에서는 첫 번째로 전이 금속 칼코겐 화합물인 텅스텐 다이셀레나이드 기반의 전계 효과 트랜지스터를 제작, 해당 전자 소자에 양성자 빔이 미치는 영향을 전기적, 물리화학적 특성 변화를 통해 논하였습니다. 전이 금속 칼코겐 화합물은 높은 이동도, 높은 온/오프 전류비, 두께에 따라 변화하는 밴드갭 등의 전계

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효과 트랜지스터로 활용되기 좋은 특성을 가져, 차세대 소자 물질로 각광받고 있는 물질입니다. 그 중 텅스텐 다이셀레나이드는 다른 전이 금속 칼코겐 화합물과는 차별적으로 두께에 따라 전자 소자의 캐리어 유형이 변화하는 독특한 특성을 가집니다. 본 학위 논문에서는 전자 축적 영역과 정공 축적 영역 모두에 대한 양성자 효과를 살펴보기 위하여 양극성 유형의 트랜지스터를 설계 및 제작하였습니다. 제작된 소자에 다양한 선량 조건에서 10 MeV의 고에너지 양성자 빔을 조사하고 조사 전 후의 전기적 및 물리화학적 특성을 체계적으로 비교하였습니다. 이 때, 물리화학적 특성은 측정한계 내에서 변하지 않았으나 전류 레벨, 문턱전압 등의 전기적 특성은 조사 선량에 따라 변화하였습니다. 이러한 변화는 양성자 조사 유도 트랩(양성 벌크 트랩 및 음성 계면 트랩 상태)으로 설명되었습니다.

두 번째로, 유기-무기 할로겐화 납 페로브스카이트를 두 가지 방법으로 합성한 후 합성법에 따른 양성자 빔에 대한 경도를 논하고자 합니다. 이 때 사용된 합성법은 기계 화학 합성법과 순간 증발 합성법이며 두 방법 모두 건식 합성법이기 때문에 대표적인 페로브스카이트 합성법 중 하나인 스핀 코팅 기법에 비해 독성이 적다는 장점이 있습니다. 합성된 물질은 다양한 조사 선량의 고에너지 양성자 빔에 노출되었으며, 순간 증발 합성법과 달리 기계 화학 합성법은 높은 조사 선량에서도 물리 화학적 성질이 크게 변화하지 않았습니다. 이는 기계 화학 합성법으로 얻어진 페로브스카이트가 더 적은 격자 결함을

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가지고 있어 물질 내 결합력이 더 강하기 때문으로 설명되었습니다.

주요어 : 양성자 빔, 고에너지 방사선, 전이금속 칼코겐 화합물, 유무기 페로브스카이트

학번: 2016-20297

Publication Lists (First Authors)

- Jiwon Shin, Kyungjune Cho, Tae-Young Kim, Jinsu Pak, Jae-Keun Kim, Woocheol Lee, Jaeyoung Kim, Seungjun Chung, Woong-Ki Hong and Takhee Lee, "Dose-dependent effect of proton irradiation on electrical properties of WSe2 ambipolar field effect transistors", Nanoscale, 11, 13961 (2019).
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Publication Lists (Co-authors)

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